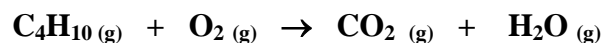
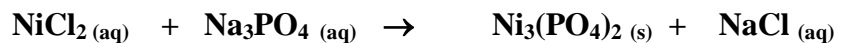


Unit #6 Practice Blackboard Insert

1. (2 pts) How many grams of aluminum carbonate are in 0.655 moles of the compound?
2. (3 pts) Calculate the % composition of Cobalt (II) Nitrate..
3. (3 pts) What is the empirical formula of a substance that contains 4.027 grams potassium, 2.678 grams chromium, and 3.296 grams oxygen?
4. (2 pts) Balance the following equations:



Extra Credit (1 pt):

Convert name to formula or visa versa ... from the list of names in unit 5