

CEM 090 – More Reactions To Balance

1.  $\text{C}_5\text{H}_{11}\text{OH} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
2.  $\text{CaH}_2 + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2$
3.  $\text{ZnS} + \text{O}_2 \rightarrow \text{ZnO} + \text{SO}_2$
4.  $\text{CH}_3\text{NO}_2 + \text{Cl}_2 \rightarrow \text{CCl}_3\text{NO}_2 + \text{HCl}$
5.  $\text{C}_6\text{H}_{12}\text{O}_6 \rightarrow \text{C}_2\text{H}_5\text{OH} + \text{CO}_2$
6.  $\text{Al} + \text{Cr}_2\text{O}_3 \rightarrow \text{Al}_2\text{O}_3 + \text{Cr}$
7.  $\text{CS}_2 + \text{Cl}_2 \rightarrow \text{CCl}_4 + \text{S}_2\text{Cl}_2$  (This disulfur dichloride ... OK as written)
8.  $\text{C} + \text{SO}_2 \rightarrow \text{CS}_2 + \text{CO}$
9.  $\text{C}_{10}\text{H}_8 + \text{O}_2 \rightarrow \text{C}_8\text{H}_4\text{O}_3 + \text{CO}_2 + \text{H}_2\text{O}$
10.  $(\text{NH}_4)_2\text{S}_2\text{O}_8 + \text{H}_2\text{O} \rightarrow \text{NH}_4\text{HSO}_4 + \text{H}_2\text{O}_2$
11.  $\text{NaF} + \text{CaO} + \text{H}_2\text{O} \rightarrow \text{CaF}_2 + \text{NaOH}$
12.  $\text{Na} + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2$
13.  $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$
14.  $\text{Zn} + \text{Cu(NO}_3)_2 \rightarrow \text{Cu} + \text{Zn(NO}_3)_2$
15.  $\text{FeCl}_3 + \text{KOH} \rightarrow \text{Fe(OH)}_3 + \text{KCl}$